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[2] (b) Marit arranged the values she found in Table 12 into an energy cycle. C2H4(g) + H2(g) C2H6(g) 3 O 2 2CO2(g) + 3H2O - 1 4 1 1 k J m o l - 1 - 2 8 6 k J m o l - 1 - 1 5 6 0 k J m o l - 1 (hydrogenation) (l) Calculate the value for the enthalpy of hydrogenation of ethene from the energy cycle.....

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8 20 2 2 2 2 o correct reactants and products; [2] correct coefficients; M2 can only be scored if M1 correct. (b) Local pollutant: carbon monoxide/CO / volatile organics/VOCs / nitrogen oxide/NO / (unburnt) hydrocarbons; Do not accept methane/CH 4, ethane/C 2 H 6, propane/C 3 H 8 or butane/C 4 H 10. Global pollutant: nitrogen oxide/NO / carbon ...

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