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Chapter 19 Mastering Chemistry Homework Spontaneity and

Reversibility: Part A For a single substance at atmospheric

pressure, classify the following as describing a spontaneous process, a nonspontaneous process, or an equilibrium system.

Part B Determine whether each of these processes is reversible or irreversible.

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Solution: Chapter 19 Electric Charges, Forces, and Fields Q.83GP. At the moment, the number of electrons in your body is essentially the same as the number of protons, giving you a net charge of zero. Suppose, however, that this balance of charges is

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off by 1% in both you and your friend, who is 1 meter away.

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Chemistry Chapter 19 Answers Chemical Thermodynamics

Example 9.2 The element mercury, Hg, is a silvery liquid at room temperature. The normal freezing point of mercury is -38.9°C , and its molar enthalpy of fusion is $\Delta H_{\text{fusion}} = 2.29 \text{ kJ/mol}$.

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