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methanol has = 791
kg/m3 and a large
vapor pressure of
13,400 Pa. Then the
manometer rise h is
given by pLaPaz pvap
64400 13400 methanol
g h (791)(9.81) h Solve
for

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...
Note: Unit cancellations are not shown in this solution.
a.) Situation: Pressure varies with elevation. z= 6:8ft. Find: Pressure change (kPa). Properties: ˆ= 90 lbm=ft3. Solution: Convert density to units of kg/m3: ˆ= 90 lbm ft 3 35:315ft3 m 1:0kg 2:2046 lbm =

For Fluid

1442 kg Calculate the pressure change: p= ˆg z= 1442kg m3 9:81m s2 6:8ft 1:0 m 3 ...

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distributed load at the base of the dam is wb = rwghb = 11000 kg > m3 219.81 m > s2 219 m211 m2 = 88.291103 2 N ...

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probability distribution,
$f(v) = f(v\ 1, v\ 2, v\ 3)$, of
molecular velocities in
a gas flow at a point in
space with average
velocity u is given by
(1.1). a) Verify that u is
the average molecular
velocity, and
determine the standard
deviations ($\sigma\ 1$,

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