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From partial molar results (eq. (8.15)), the free energy of the solution is  $G_{\text{soln}} = n_A G_{A0} + n_B G_{B0} + RT [n_A \ln a_A + n_B \ln a_B] + H_{\text{mix}} + n_B L$  Because  $G_{A0}$  and  $G_{B0}$  are for constants that do not depend on subsequent increase in temperature, we need to solve for an increase in  $G_{\text{soln}}$  using  $\Delta G_{\text{soln}} = RT [n_A \ln a_A + n_B \ln a_B] + H_{\text{mix}} + n_B L$  ...

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David R. Gaskell received a B.Sc. in Metallurgy and Technical Chemistry from the University of Glasgow and a Ph.D. from McMaster University. Dr. Gaskell's first faculty position was at the University of Pennsylvania, Philadelphia, where he taught from 1967 to 1982 in Metallurgy and Materials Science.

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Elizabeth Cleghorn Gaskell (née Stevenson; 29 September 1810 - 12 November 1865), often referred to as Mrs Gaskell, was an English novelist, biographer and short story writer. Her novels offer a detailed portrait of the lives of many strata of Victorian society, including the very poor, and are of interest to social historians as well as readers of literature.

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